

# GCSE

## **Chemistry A**

Unit A173/01: Module C7 (Foundation Tier)

General Certificate of Secondary Education

### Mark Scheme for June 2017

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This mark scheme is published as an aid to teachers and students, to indicate the requirements of the examination. It shows the basis on which marks were awarded by examiners. It does not indicate the details of the discussions which took place at an examiners' meeting before marking commenced.

All examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes should be read in conjunction with the published question papers and the report on the examination.

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#### Annotations

Used in the detailed Mark Scheme:

Annotation	Meaning
/	alternative and acceptable answers for the same marking point
(1)	separates marking points
not/reject	answers which are not worthy of credit
ignore	statements which are irrelevant - applies to neutral answers
allow/accept	answers that can be accepted
(words)	words which are not essential to gain credit
words	underlined words must be present in answer to score a mark
ecf	error carried forward
AW/owtte	credit alternative wording / or words to that effect
ORA	or reverse argument

Available in RM Assessor to annotate scripts:

?	indicate uncertainty or ambiguity
BOD	benefit of doubt
CON	contradiction
×	incorrect response
ECF	error carried forward
	draw attention to particular part of candidate's response
NBOD	no benefit of doubt
R	reject
	correct response

L1 , L2 , L3	draw attention to particular part of candidate's response
	information omitted
?	indicate uncertainty or ambiguity
BOD	benefit of doubt
CON	contradiction
×	incorrect response
ECF	error carried forward
$\bigcirc$	draw attention to particular part of candidate's response
	draw attention to particular part of candidate's response
	draw attention to particular part of candidate's response
NBOD	no benefit of doubt
R	reject
✓	correct response
3	draw attention to particular part of candidate's response
	information omitted

#### Subject-specific Marking Instructions

Accept any clear, unambiguous response (including mis-spellings of scientific terms if they are *phonetically* correct, but always check the guidance column for exclusions).

Crossed out answers should be considered only if no other response has been made. When marking crossed out responses, accept correct answers which are clear and unambiguous.

e.g. for a one-mark question where ticks in the third <u>and</u> fourth boxes are required for the mark:



The list principle:

If a list of responses greater than the number requested is given, work through the list from the beginning. Award one mark for each correct response, ignore any neutral response, and deduct one mark for any incorrect response, e.g. one which has an error of science. If the number of incorrect responses is equal to or greater than the number of correct responses, no marks are awarded. A neutral response is correct but irrelevant to the question.

Marking method for tick-box questions:

If there is a set of boxes, some of which should be ticked and others left empty, then judge the entire set of boxes. If there is at least one tick, ignore crosses and other markings. If there are no ticks, accept clear, unambiguous indications, e.g. shading or crosses. Credit should be given according to the instructions given in the guidance column for the question. If more boxes are ticked than there are correct answers, then deduct one mark for each additional tick. Candidates cannot score less than zero marks.

e.g. if a question requires candidates to identify cities in England:



the second and fourth boxes should have ticks (or other clear indication of choice) and the first and third <u>should be blank</u> (or have indication of choice crossed out).

Edinburgh										
Manchester		×								
Paris										
Southampton		×								
Score:	2	2	1	1	1	1	0	0	0	NR

For answers marked by levels of response:

- i. Read through the whole answer from start to finish
- ii. Decide the level that best fits the answer match the quality of the answer to the closest level descriptor
- iii. To determine the mark within the level, consider the following:

Descriptor	Award mark
A good match to the level descriptor	The higher mark in the level
Just matches the level descriptor	The lower mark in the level

iv. Use the L1, L2, L3 annotations in RM Assessor to show your decision; do not use ticks.

Quality of Written Communication skills assessed in 6-mark extended writing questions include:

- appropriate use of correct scientific terms
- spelling, punctuation and grammar
- developing a structured, persuasive argument
- selecting and using evidence to support an argument
- considering different sides of a debate in a balanced way
- logical sequencing.

Que	stion	Answer		Marks	Guidance
1	(a)	stage needed	equipment	3	(1) for each correct line.
			burette		If more than one line either to or from a box, that box does not 'count'.
		measure mass of	volumetric		
		solid	flask		
		dissolve solid in	beaker and		
		water and stir	glass rod		
		make solution up	balance		
		to exactly 250 cm <sup>3</sup>			
			thermometer		

Question	Answer	Marks	Guidance
Question 1 b	Answer         [Level 3]         Makes statements about sampling, storing and handling and making solutions.         Quality of written communication does not impede communication of the science at this level.         (5 – 6 marks)         [Level 2]         Makes a statement about two different aspects of the procedure.         Quality of written communication partly impedes communication of the science at this level.         (3 – 4 marks)         [Level 1]         Makes a basic statement about one aspect of the procedure.         Quality of written communication impedes communication of the science at this level.         (1 – 2 marks)         [Level 0]         Insufficient or irrelevant science. Answer not worthy of credit.         (0 marks)	Marks 6	Guidance         This question is targeted at grades up to D         Indicative scientific points may include: Level         3 (Accuracy)         Level 2 and 3 (Use of equipment)         Sampling         random sampling/ spread of samples         idea of different times during the day         regular spacing of times         more than one sample at each time         idea of taking lots of samples to take a mean value         Storing and Handling         using containers for different tablets         idea of keeping samples separate / not mixing up         labels on samples         not contaminating / wear gloves / make sure         equipment is clean         separate glassware/equipment         Making solutions         control of volume or amount of water/ total volume or amount of solution         control of number/amount of tablets (used to make solution)         idea of not contaminating glassware / using clean equipment.         Allow 'repeat' idea for L1 (1) mark as evidence of partial engagement with science
			engagement with science Use the L1, L2, L3 annotations in RM Assessor; do not use ticks.

Que	Question		Answer	Marks	Guidance
1	(C)		sodium chloride ; (1)	2	
			$CO_2; (1)$		
1		ii	reacts with/ neutralises (hydrochloric) acid/HCI ; (1)	2	
			Carbon dioxide/CO2 quoted as the product; (1)		Do not allow second marking point if other products are
			Total	10	listed
			Total	13	
Que	stion	1	Answer	Marks	Guidance
2	(a)	(i)	Any 2 from:	2	Alternatives to first point
					results not concordant/consistent/repeatable/ values are
			Range is wide/ results vary ;		quite far away from each other / fluctuates too much/
					there is a range
			Identifies range 8.0 / 19.0 to 27 ;		
					Ignore reliable
			19.0/ result 4 is an outlier ;		
					outlier must be identified
			Rough reading should be above accurate ;		
	(ii)		a ninette	1	
	()			•	
	(b)	(i)	21.0 20.5 21.5	1	All three needed
	(ii)		21(.0) (2 marks)	2	Allow ecf on three ringed values from (b) (i) for (2) marks
			For (1) mark		Allow 22.2 / 22(.0) for (1) mark
			adds ringed values ; (1)		(includes all five or last four in average)
	(c)		concentration is 4% ; (1)	2	
1			which is too low / below 5 % / different to 5% / best		ignore different alone
			estimate/it should be 25.0; (1)		
			Total	8	

Question	Answer	Marks	Guidance
3	[] 0/0] 3]	6	This question is targeted at grades up to C
	Makes statement about a similarity and a difference to include activation energy and energy change of		Indicative scientific points may include:
	reaction.		Similarities
	Quality of written communication does not impede communication of the science at this level. (5 – 6 marks)		<ul> <li>energy given out when hydrogen burns is the same as the energy taken in when it forms.</li> <li>Both reactions involve hydrogen, oxygen and water</li> <li>the energy of water OR hydrogen and oxygen is the same in both reactions</li> </ul>
	[Level 2]		
	Gives a similarity and a difference between the two diagrams. OR		<ul> <li>Differences</li> <li>making has larger activation energy / burning has smaller</li> </ul>
	Gives differences or similarities between the two diagrams.		<ul><li>making endothermic</li><li>burning exothermic</li></ul>
	Quality of written communication partly impedes communication of the science at this level. (3 – 4 marks)		<ul> <li>making energy needed/taken in</li> <li>burning energy given out</li> <li><u>energy</u> change of reaction is in different direction/opposite.</li> </ul>
	<b>[Level 1]</b> Makes a statement about a similarity or a difference between the diagrams.		Allow 'energy decreases when burning hydrogen and increases when making hydrogen'
	Quality of written communication impedes communication of the science at this level. (1 – 2 marks)	– 2 marks) If activation energy po omitted, consider QW	If activation energy point is made but 'activation' is omitted, consider QWC impeded <b>e,g</b> , 'larger amount of
	[Level 0]		energy <u>needed/ t aken in</u> for making hydrogen
	Insufficient or irrelevant science. Answer not worthy of credit. (0 marks)		Use the L1, L2, L3 annotations in RM Assessor; do not
		6	
		U	

Que	Question		Answer		Marks	Guidance	
4	(a)	(i)	C <sub>10</sub> ; (1)			2	
			H <sub>22</sub> ; (1)				
		(ii)		true	false	3	all correct (3)
			decane has a higher relative formula mass than octane				3 correct (2)
			both molecules contain double bonds				
			both molecules are hydrocarbons				
			both molecules give off carbon dioxide gas when they burn				
	(b)	(i)	H H H H $H - C - C - C - C - H$ $H - C - C - C - C - H$ $H H H H$ $H H H$ $H H$ $(1)$			2	NO ecf on the name All bonds and hydrogen atoms should be shown
					Total	7	

Que	stion	Answer	Marks	Guidance	
5	(a)	reversible;	1		
	(b)	<i>any 2 from:</i> does not all react / so all ethene is used ;	2	Ignore 'to use it again'	
		to avoid waste idea / saves resources / crude oil / sustainable;		<b>Ignore</b> atom economy / saves energy ; Allow 'no waste' Allow saves <u>fossil</u> fuel (crude oil)	
		to increase yield / to make more ethanol ;			
	(c)	<i>Any 2 from :</i> high temperature ;	2	Allow 'optimum' temperature	
		high pressure;			
		catalyst;			
	(d)	cools / goes into cooler ;	2		
		condenses ;			
	(e)	fermentation of sugar	2		
		using genetically modified			
		in the Haber process using gas			
		chromatography			
		by titration			
		Total	9		

Question	Answer	Marks	Guidance
Question 6	Answer         [Level 3]         Gives a benefit and risk and makes a statement about the data.         Quality of written communication does not impede communication of the science at this level.         (5 – 6 marks)         [Level 2]         Gives a benefit or risk and makes a statement about the data.         OR gives a benefit and a risk without clear reference to data.         Quality of written communication partly impedes communication of the science at this level.         (3 – 4 marks)         [Level 1]         Makes a statement about risk, benefit or data.         Quality of written communication impedes communication of the science at this level.         (1 – 2 marks)         [Level 0]         Insufficient or irrelevant science. Answer not worthy of credit.         (0 marks)	Marks 6	Guidance         This question is targeted at grades up to C         Indicative scientific points may include: Risks         methane and/or octane may be too/very flammable / cause a fire risk         gas can escape/difficult to store         Benefits         liquid fuels/heptane/decane easier to carry or store as fuel / less likely to leak / can see leaks easily ORA         ignites/burns in cold (weather)         liquid fuels/decane/octane easy to carry/store as fuel / less likely to leak / can see/deal with leaks easily         compares energy output from 2 fuels         decane gives out the most energy when burned.         Data         methane is a gas / heptane and/or decane/octane are liquids         methane has the lowest flashpoint / decane has the highest flashpoint / methane/octane ignites/burns easily / flammable /decane does not ignite/burn easily         decane is less of a fire risk         methane gives out the least energy when burned         Use the L1, L2, L3 annotations in RM Assessor; do not use ticks.
		6	

Question			Answer		Marks	Guidance
7	(a)		old process has a waste (product)/ sodium sulfit new process has no waste (product) / all produc useful ; (1)	te ; (1) cts are	2	Ignore statements about yield Ignore ' <i>less</i> waste'
	(b)		any 3 from:		3	ignore all the atoms are used
			higher yield ; higher atom economy ; does not have any waste (products) / only by-pr all products useful ; (waste from older process) toxic/harmful ;	roducts /	5	ignore statements about energy
			needs less/fewer raw materials/ does not use si acid/ sodium hydroxide	ulfuric		"releases no harmful by-products" is this last point only [it may be releasing other by-products, but none of them are harmful]
	(c)	i	using renewable raw materials		2	
			using a higher temperature and pressure			
			finding more uses for phenol			
			finding ways to increase the yield of phenol			

	(d)	ii	to make sure that other scientists do no	ot take crec their	lit for work	2	
			to reduce the safety risks during their experime so that other scientists can check their data so that they can discuss their conclusions	ents a			
	to stop other scientists from working on the same idea						
	(d)			adv	disadv	2	All correct (2)
			enzymes speed up reactions				2 or 3 correct (1)
			reactions with enzymes can work at a lower temperature				
			enzymes only work in narrow ranges of pH and temperature	nes only work in narrow ranges of pH			
			enzymes can be denatured				
					Total	11	

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